## THE ISOLATION OF VINCANINE AND VINCANIDINE

## FROM THF ROOTS OF Vinca erecta

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The roots of the plant  $\underline{V}$  erecta growing in Central Asia form the raw material for the isolation of vincanine [1] and vincanidine [2].

When vincanine is produced by a published method [3], the vincanidine remains almost completely on the cation-exchange resin, and its desorption requires the consumption of a large amount of solvent and time.

We have developed a method for isolating these alkaloids from the roots of  $\underline{V}$ ,  $\underline{erecta}$  in which it is possible to obtain vincanidine in good yield and to increase the yield of vincanine by  $\underline{10\%}$  as compared with the existing method.

The comminuted roots collected in 1972 in the Osh oblast (5 kg) were charged into five extractors and extracted with 1%  $H_2SO_4$  by the battery extraction method at the rate of 226 liters/h·m² (consumption 4 liters/h). Each 10 liters of extract was made alkaline with conc.  $NH_4OH$  to pH 8-8.5, and the alkaloids were extracted with chloroform. The aqueous extract was checked for the absence of alkaloids by thin-layer chromatography on silica gel [4]. A total of 50 liters of extract was obtained which was treated with 30 liters of chloroform.

The combined chloroform extracts were evaporated to 5 liters, and the phenolic alkaloids were extracted with 5% NaOH.

The alkaline solution was made acid with 10% H<sub>2</sub>SO<sub>4</sub>, washed once with ether, and was made alkaline again with NH<sub>4</sub>OH to pH 8-8.5 and the phenolic alkaloids were extracted with 20 liters of ether. When the ether was concentrated, crystals of vincanidine precipitated – 17.5 g, or 70% of the amount in the raw material.

The nonphenolic alkaloids from the chloroform were transferred into 10% H<sub>2</sub>SO<sub>4</sub>, and this solution was made alkaline to pH 8-8.5 and extracted with 10 liters of benzene. Evaporation of the benzene solution gave 25.5 g of vincanine, or 85% of the amount in the raw material.

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